TWO POLYACETYLENIC PHYTOALEXINS FROM ARCTIUM LAPPA*

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Key Word Index—Arctium lappa; Compositae; burdock; stress metabolites; phytoalexins; polyacetylenes.

Abstract—Treatment of sliced burdock root tissue with copper (II) sulphate stimulated phytoalexin formation. Two were isolated and characterized as (S)-12,13-epoxy-2,4,6,8,10-tridecapentayne and 1-tridecen-3,5,7,9,11-pentayne by spectroscopic methods and some chemical transformations.

INTRODUCTION

Although the Compositae is one of the biggest plant families and has been a rich source of secondary metabolites, few phytoalexins have been reported from this family [2]. Two polyacetylenes, safynol and dehydrosafynol [3, 4], and a coumarin ayapin [5] were recorded as phytoalexins of safflower (Carthamus tinctorius) and sunflower (Helianthus annuus), respectively. Recently, we reported two sesquiterpenoid phytoalexins, lettucenin A and costunolide from lettuce (Lactuca sativa) [6]. In a continuation of our search for composite phytoalexins we examined burdock Arctium lappa (gobo in Japanese), whose root is used as a food and a folk medicine in Japan. The present report describes the isolation and structure elucidation of two burdock phytoalexins 1 and 2 as (S)-12,13-epoxy-2,4,6,8,10-tridecapentayne and 1-tridecen-3,5,7,9,11-pentayne, respectively.

RESULTS AND DISCUSSION

Treatment of sliced burdock roots with phytopathogenic bacteria, copper sulphate (CuSO₄), or UV irradiation induced the production of several antifungal compounds as evidenced by TLC bioassay. For preparative work, the slices were treated with 0.5% aq. CuSO₄ and then incubated for 3 days. The acetone extract from the incubated slices gave two extremely unstable compounds 1 and 2 after repeated chromatography on silica gel. The compounds 1 and 2 completely inhibited the conidial germination of Bipolaris leersiae at concentrations of 0.25 and 100 ppm, respectively.

The first compound 1, $C_{13}H_6O$, $[\alpha]_D = +52.8^\circ$ (CHCl₃), showed the UV spectrum characteristic of conjugated pentaynes [7] (see Experimental). Its ¹H NMR spectrum indicated the presence of one methyl group attached to a triple bond (δ 2.00, 3H, singlet) and one terminal epoxide ring (δ 2.96, 2H, d-like, J = 3 Hz; δ 3.43, 1H, t-like, J = 3 Hz). Catalytic hydrogenation of 1

The second compound 2, C₁₃H₆, showed spectral data (see Experimental) consistent with 1-tridecen-3,5,7,9,11-pentayne [7]. This compound has been found in many Compositae plants in very low concentrations [7]. Some species of Arctium [9] also contain 2. Washino et al. reported recently eight polyacetylenes from A. lappa [10].

$$Me(C = C)_5 = CH_2$$
 $Me(C = C)_5 CH = CH_2$

1

2

4a 2S, R = H

4b 2R, R = H

5a 2S, R = (R)-MTPA

5b 2R, R = (R)-MTPA

over PtO₂ afforded 1-tridecanol confirming 1 to be 12,13epoxy-2,4,6,8,10-tridecapentayne. The absolute configuration of 1 was next examined. Catalytic hydrogenation of 1 using Pd on BaSO₄ gave a saturated epoxide 3b, which gave a 2-tridecanol 4b on lithium aluminium hydride reduction. The absolute configuration of (+)-1,2epoxytridecane (3a) is known to be R [8]. Since the reported optical rotation of 3a was so small [8] and the available amounts of the epoxide 3b were limited, the alcohol 4b was converted to a MTPA-ester 5b (see Experimental) to compare it directly with the MTPA-ester 5a derived from 4a. Comparison of the both MTPAesters (5a and 5b) by HPLC and ¹H NMR spectroscopy indicated that 5b was not identical to 5a. Consequently, 1 should have the (12S)-configuration. Although 12,13epoxy-2,4,6,8,10-tridecapentayne is a known polyacetylene [7] of composite plants, there has been no report on its absolute configuration.

^{*}Part 5 in the series 'Studies on Stress Metabolites'. For part 4 see ref. [1].

However, they mentioned the absence of 2 in the roots of the plant. We found that the amounts of 2 in the peeled control root tissue were negligible but increased 30 times in the peeled CuSO₄-treated root tissue. Interestingly, the concentration of 2 in the untreated epidermal root tissue was much higher than that of the inner control root tissue, suggesting some role for 2 as a plant-defence mechanism.

Acetylenic phytoalexins have been also reported from other plant families; furanoacetylenes from *Vicia* and *Lens* species (Leguminosae) [11], falcarindiol from carrot (Umbelliferae) [12] and tomato (Solanaceae) [13].

EXPERIMENTAL

¹H (100 or 400 MHz) NMR spectra and optical rotation were recorded in CDCl₃ and in CHCl₃, respectively. MS were measured with a direct inlet system at 70 eV.

TLC bioassay. Devloped silica gel sheets (Merck, Kieselgel 60 F₂₅₄; ether) were air-dried, sprayed with a dense conidial suspension of *Bipolaris leersiae* in potato-glucose medium, and incubated in a moist box at 25° for 2 days [1].

Induction and isolation of (S)-12,13-epoxy-2,4,6,8,10tridecapentayne (1) and 1-tridecen-3,5,7,9,11-pentayne (2). Burdock roots, A. lappa L. cv Shirohada-sakigake, (30 kg) were washed, the epidermal tissue removed and cut into slices 2 mm thick. The slices were kept in moist boxes at 25° overnight and then treated with 0.5 % aq. CuSO₄. After being incubated at 25° for 3 days, the slices, which became brown, were freeze-dried (2.7 kg) and extracted \times 3 with Me₂CO. The combined extract was concd to 2.01 and kept at 5° until used to avoid decomposition of sensitive active compounds. A one-tenth portion of the extract was evapd to leave 5.1 g of residue, which on silica gel CC using Et₂O as eluent gave 6 bioactive fractions. The least polar active fraction (1.2 g) was further sepd by sequential chromatography using silica gel (CH2Cl2-MeOH, 49:1), µ-Porasil (n-hexane-CH₂Cl₂, 2:1), and μ -Porasil (n-hexane) columns to give 1 (52 mg) and 2 (2 mg).

(S)-12,13-Epoxy-2,4,6,8,10-tridecapentayne (1). Pale yellow crystals, decomposing immediate!y and becoming violet to black in colour; high resolution MS, m/z 178.04375 [M]⁺ (calc. for $C_{13}H_6O$: 178.04357); EIMS, m/z (rel. int.): 178 (82), 162 (55), 148 (100), 122 (84); $[\alpha]_D$, +52.8°; UV λ_{max} (MeOH) nm: 377 (\$\alpha\$00), 350 (1700), 326 (1700), 285 (1100), 264 (144000), 251 (103000), 238 (49000), 226 (19000); IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 2100 and 2200 (-C=C-), 860 (epoxide); ¹H NMR: δ 2.00 (3H, s), 2.96 (2H, d-like, J = 3 Hz), 3.43 (1H, t-like, J = 3 Hz).

Catalytic hydrogenation of 1. (i) The epoxide 1 (5 mg) was hydrogenated (PtO₂ 2 mg, EtOH 3 ml, 1 hr) and the product purified by silica gel CC with CH₂Cl₂-MeOH (49:1) yielding 1-tridecanol; EIMS, m/z (rel. int.): 182 (2, $[M-H_2O]^+$, 154 (4), 125 (6), 111 (13), 97 (32), 83 (50), 69 (65), 55 (83), 43 (100); ¹H NMR: δ 0.88 (3H, t, J = 7 Hz), 1.2 (22H, br s), 3.64 (2H, t, J = 6 Hz). (ii) The epoxide 1 (5 mg) was hydrogenated (5 % Pd on BaSO₄ 9 mg, n-hexane, 1 hr) and the products sepd by silica gel CC with CH₂Cl₂-CCl₄ (1:2) to give a satd epoxide 3b (1 mg).

LiAlH₄ reduction of 1,2-epoxytridecanes (3a and 3b). (i) A mixture of (R)-(+)-1,2-epoxytridecane (3a) (1.0 g) [8] and LiAlH₄ (2.4 g) in Et₂O was refluxed for 16 hr. After usual work-up 0.91 g of (S)-(+)-2-tridecanol (4a) was obtained. (ii) The

saturated epoxide 3b (1 mg) from 1 was treated similarly with LiAlH₄ to give an enantiomeric alcohol 4b (0.7 mg).

Preparation of MTPA esters. (i) (S)-(+)-2-tridecanol (4a, 100 mg) was converted to the MTPA ester 5a (158 mg) of (R)-(+)-α-methoxy-α-(trifluoromethyl) phenylacetic acid according to the method of ref. [14]. ¹H NMR: δ 3.57 (3H, –OMe, with ca 1 Hz long-range coupling). (ii) The tridecanol (4b) from 1 was converted similarly to a MTPA ester 5b (1.2 mg). ¹H NMR: δ 3.55 (3H, –OMe, with ca 1 Hz long-range coupling). (iii) Racemic 2-tridecanol 4c (126 mg) was converted similarly to the corresponding diasteromeric mixture of MTPA esters 5c (155 mg). ¹H NMR: δ 3.55 and 3.57 (–OMe).

HPLC analysis of MTPA esters, 5a, 5b, and 5c. HPLC [μ -Porasil, n-hexane—CH₂Cl₂ (4:1), 1 ml/min] of 5a, 5b, 5c showed R_1 s of 14.6, 16.2, 14.6 and 16.2 min, respectively.

1-Tridecen-3,5,7,9,11-pentayne (2). Highly unstable yellow crystals, decomposing rapidly; high resolution MS, m/z 162.0454 [M]⁺ (calc. for C₁₃H₆: 162.0470); EIMS, m/z (rel. int.): 162 (100), 136 (32), 110 (29); UV λ_{max} (MeOH) nm 410 (ε1500), 378 (5500), 352 (5800), 330 (3100), 308 (1500), 285 (58 300), 269 (63 500), 264 (72 900), 257 (55 700), 252 (sh, 46 700), 245 (sh, 31 100); IR ν_{max} (CHCl₃) cm⁻¹: 3020, 2225 and 2190 (-C≡C-), 390 (-CH=CH₂); ¹H NMR: δ2.00 (3H, s, Me) 5.7–6.0 (3H, m, -CH=CH₂).

HPLC analysis of 2 in different tissues. Each Me₂CO extract from respective tissue of single burdock root was submitted to HPLC (μ-Porasil, n-hexane) and peak areas corresponding to 2 (R₁ 7.5 min) were compared. Relative amounts of 2 from peeled, peeled and CuSO₄-treated root tissues, and root epidermal tissue were in the ratio of 1, 30, and 4, respectively.

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